

## **THE PHOSPHORIC ACID INFLUENCE ON THE DEPOLLUTION OF THE WASTE WATERS ORIGINATED IN THE PRODUCTION OF KRAFT PULP**

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**ABSTRACT:** The obtaining of kraft pulp from softwood through the leaching of lignine in a highly alkaline environment leads to the pollution of the environment because of the existence of the alkalilignin and the tiolignin in the waste waters.

**KEYWORDS:** kraft pulp, wastewaters, phosphoric acid, depollution

### **INTRODUCTION**

Sulphate waste waters resulted in the technological stages of the wood grinding, washing, screening and the recovery of the sodium salts are characterized by a highly polluting chemical composition high biodegradable.

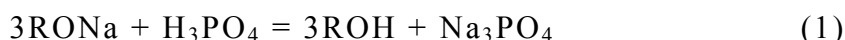
*Table 1. The chemical composition of sulphate waste waters*

<b>The label of the qualitative indicator</b>	<b>Absolute value</b>
Fixed residue at $100 \pm 5^{\circ}\text{C}$ , %	80
Content of the organic substances from fixed waste, %	90
pH	12
Chemical oxygen demand expressed as CODMn, mg/L	80000
Temperature, $^{\circ}\text{C}$	20
Colour – brownish rendered through the photometric absorption, %	6,5

The chemical characterization of the waste waters indicates a highly polluted environment.

## MATERIALS AND METHODS

The research that was carried out had in view the alkalilignin flocculation in a phosphoric acid medium. There is the general equation of the process:



The independent variables of the system were: the quantity of  $\text{H}_3\text{PO}_4$ , the temperature and time of reaction (table 2).

*Table 2*

Independent variables	The range of fluctuation ( $\Delta X = \text{cst.}$ )				
	-1,678	-1	0	1	1,678
$\text{H}_3\text{PO}_4$ (mg/100 ml water) $X_1$	3	5	7	9	11
Temperature ( $^{\circ}\text{C}$ ) $X_2$	10	15	20	25	30
Time of reaction (minutes), $X_3$	8	10	12	14	16

## RESULTS AND DISCUSSIONS

The interpretation of the results has been carried out through the particularization of the general equation of regression established for the dependent variables:

$$Y = b_0 + \sum b_i X_i + \sum b_{ij} X_i X_j + \sum b_{ii} X_{ii} \quad (2)$$

The polluting degree of the waste waters is given in figures 1, 2 and 3 through the method of particularization for CODMn, the colour index and the pH.

The discoloring of the sulphate waste waters through the lignin flocculation is achieved dependent on the  $\text{H}_3\text{PO}_4$  addition, the reduction of the CODMn reaching bellow 90% and down to 79% for that of the colour.

The discoloring of the waste waters is ended even if there is a minimum consumption of  $\text{H}_3\text{PO}_4$  of 3 mg/100 ml waste waters when the time of reaction is kept within the centred range (the real time is 12 minutes) and the reduction of the colour from blackish-brown towards yellowish corresponds to a reduction of 79%.

## CONCLUSIONS

The flocculation of lignine with  $\text{H}_3\text{PO}_4$  in the sulphate waste waters leads to:

- the discoloring from the initial blackish-brown towards yellowish colour of the waste water with a positive influence on the quality of the surface water;
- the reduction of the polluting degree through the reduction of the polluting organic components, the reduction of CCOMn reaching to 90%;

- the surplus oh  $H_3PO_4$  in the system may be used as a nutrient in the next stage of biological water treatment.

Color index, %

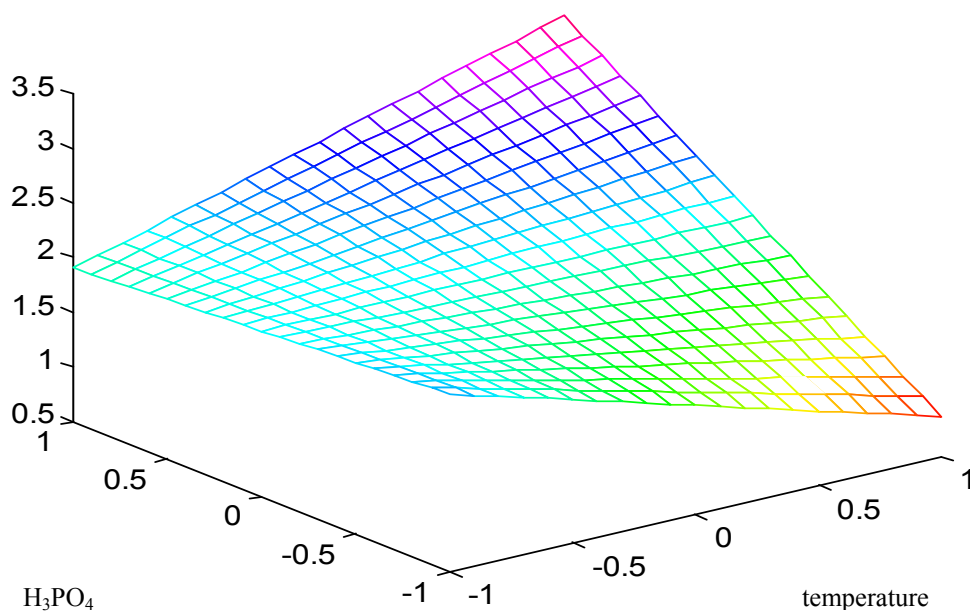


Figure 1. The colour index dependent on the  $H_3PO_4$  addition and on the temperature when the maintaining of time of reaction within the centred range

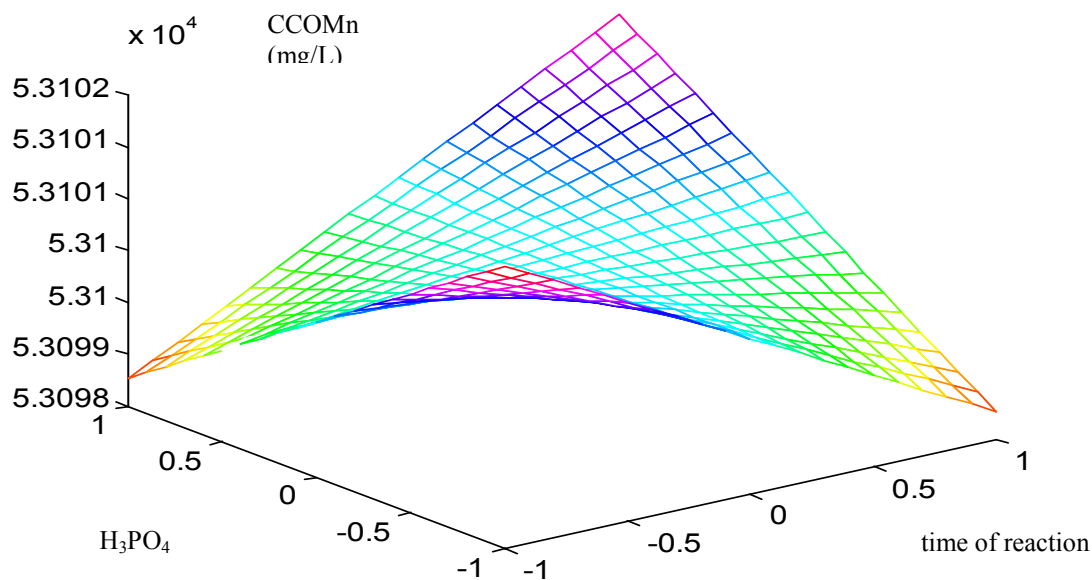


Figure 2. The CCOMn variation dependent on the  $H_3PO_4$  addition and the time of reaction

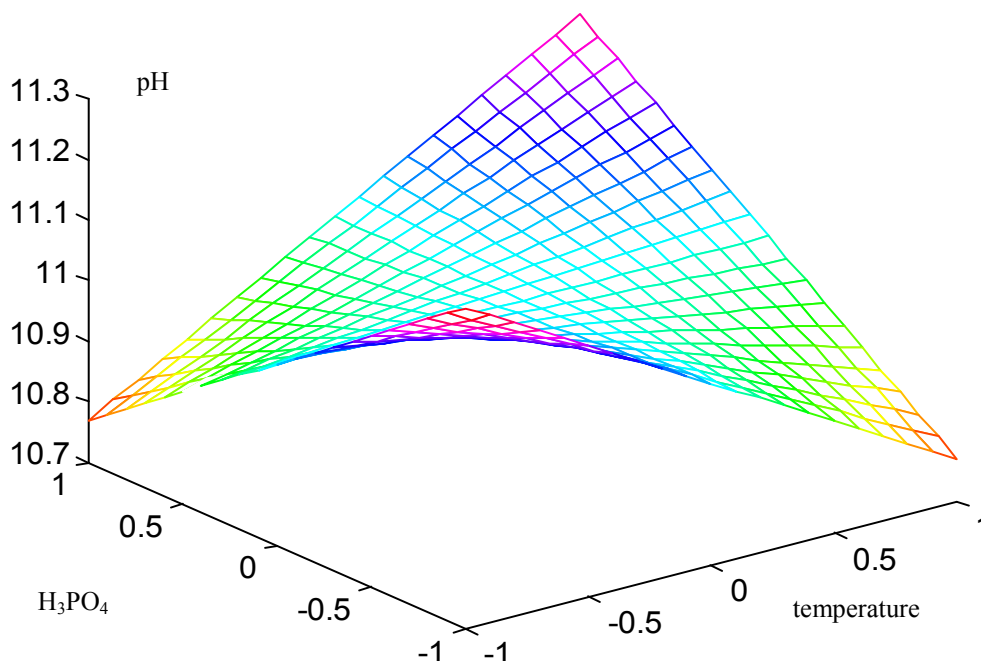


Figure 3. The pH variation dependent on the  $H_3PO_4$  addition and on the temperature

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