

$\text{Me}_4\text{NOH}(\text{SnX}_5\text{NMe}_4)_2$ (X = Cl, Br): SYNTHESIS, IR AND MÖSSBAUER STUDIES

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Abstract: Allowing Me_4NOH to react with SnX_4 (X = Cl, Br), yields $\text{Me}_4\text{NOH}(\text{SnX}_5\text{NMe}_4)_2$. Their IR and Mössbauer studies have allowed suggesting supramolecular structures, the environment around the tin center being octahedral.

Keywords: *IR, Mössbauer, pentahalotin (IV) adducts, spectroscopy, supramolecular structures*

INTRODUCTION

If many $[\text{SnCl}_5]^-$ complex-anion containing adducts [1-4] or $\text{SnCl}_3\text{OH}\cdot\text{THF}$ [5] were reported, no complex containing $[\text{OH}\dots\text{SnCl}_5]^{2-}$ have been found in the literature. We have initiated this study for understanding the behavior of Me_4NOH towards SnCl_4 , expecting an addition or a partial substitution of the chloride by OH^- and have obtained to adducts containing $[\text{SnX}_5\text{OH}]^{2-}$ ($\text{X} = \text{Cl}, \text{Br}$) what indicates the appearance of Me_4NX *in situ*. The infrared and Mössbauer studies of the compounds have been carried out and structures suggested on the basis of spectroscopic data.

EXPERIMENTAL

On allowing Me_4NOH as ethanolic solution to react with SnX_4 ($\text{X} = \text{Cl}, \text{Br}$) in benzene a white ($\text{X} = \text{Cl}$) or yellow precipitate ($\text{X} = \text{Br}$) is obtained, stirred no less than two hours and filtered. Their analytical data have allowed us to suggest the title formulae, $\text{Me}_4\text{NOH}(\text{SnX}_5\text{NMe}_4)_2$.

The reaction equation is:



Analytical data of synthesized compounds are presented in Table 1.

Table 1. Analytical data for the synthesized $\text{Me}_4\text{NOH}(\text{SnX}_5\text{NMe}_4)_2$

Compound	C		H		N		X		Sn	
	C	F	C	F	C	F	C	F	C	F
A ($\text{X} = \text{Cl}$)	17.32	17.40	4.33	4.35	5.05	5.10	42.70	42.50	28.55	28.32
B ($\text{X} = \text{Br}$)	11.28	11.34	2.82	2.91	3.29	3.33	62.67	62.54	18.59	18.70

C - % calculated; F - % found.

All the chemicals are from Aldrich Company and were used without any further purification. The infrared spectra were recorded by a PE 580 ($4000 - 200 \text{ cm}^{-1}$) or a FTIR-Nicolet ($600 - 50 \text{ cm}^{-1}$) spectrometer, at the University of Padova, Italy, the sample being as Nujol mulls while CsI or polyethylene windows were used. Mössbauer spectra were obtained as described previously [6]. Infrared data are given in cm^{-1} ; abbreviations: (vs) very strong, (s) strong, (m) medium, (sh) shoulder. Mössbauer parameters are given in $\text{mm}\cdot\text{s}^{-1}$; abbreviations: Q.S = quadruple splitting, I.S. = isomer shift, Γ = full width at half height.

RESULTS AND DISCUSSION

Let us consider the more relevant IR bands and their assignments:

(A) $\nu\text{OH} = 3545 \text{ m}$; $\delta\text{OH} = 1647 \text{ m}$; $\nu\text{SnCl}_5 = 300 \text{ s}$;

(B) $\nu\text{OH} = 3550 \text{ m}$; $\delta\text{OH} = 1650 \text{ m}$; $\nu\text{Sn-O} = 312 \text{ m}$; $\nu\text{SnBr}_5 = 200 \text{ s}$;

and their Mössbauer data:

(A) I.S. = 0.25 mm/s ; Q.S. = 0.49 mm/s ; $\Gamma = 1.07 \text{ mm/s}$;

(B) I.S. = 0.64 mm/s ; Q.S. = 0.71 mm/s ; $\Gamma = 0.93 \text{ mm/s}$.

The intensities and the frequencies of the νSnX_5 are similar to those yet reported in [7]. The I.S. values lower than those reported for SnX₆TMN₂ [8] as expected when an oxygen atom substitutes a halogen.

The presence of the absorption at 3395 cm⁻¹ on the two IR spectra are indicative of the presence of O-H...X hydrogen bond that allows to suggest while ignoring the secondary interactions, the structure reported in Figure 1. While considering O-H...X hydrogen bonds, supramolecular architectures are obtained, a string of which is reported on Figure 2.

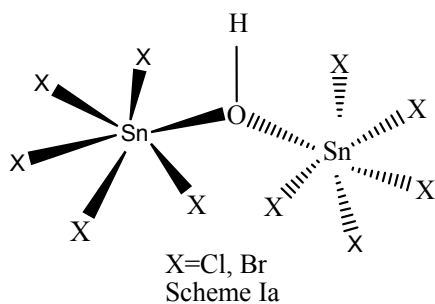


Figure 1.

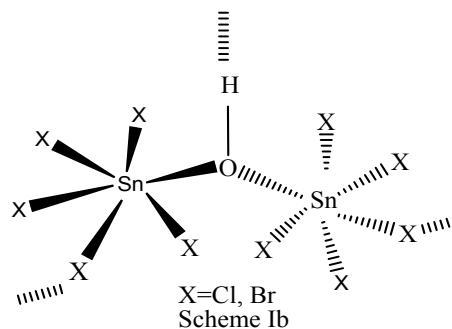


Figure 2.

CONCLUSION

The appearance in situ of Me₄NX leading to the pentahalotin (IV) anion SnX₅⁻ is noteworthy. When secondary interactions are involved as hydrogen bonds, supramolecular structures are obtained.

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